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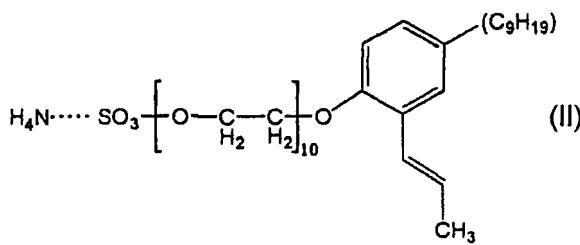
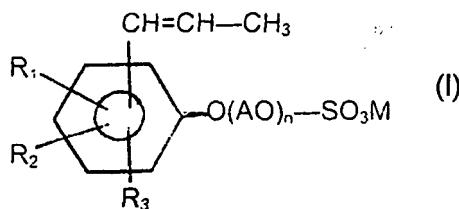
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(54) Title: WATER-WHITENING RESISTANT LATEX EMULSION PRESSURE SENSITIVE ADHESIVE AND ITS PRODUCTION



(57) Abstract: An aqueous, blush-retardant pressure sensitive adhesive (PSA) is made from an aqueous latex emulsion having an average particle size diameter of not substantially above about 100 nm and emulsified in the presence of an emulsifier consisting essentially of (I). In structure (I), R₁ is an alkyl, alkenyl, or aralkyl group containing between 6 and 18 carbon atoms, R₂ is H or R₁, R₃ is H or a propenyl group; A is an alkylene group of 2 to 4 carbon atoms, N is an integer ranging from 1 to 200, and M is an alkali metal, an ammonium ion, or an alkanolamine cation. Preferably, the emulsifier is (II). The preferred aqueous latex emulsion is prepared from a monomer mixture consisting essentially of at least one alkylacrylate having at least 4 carbon atoms in the alkyl chain, at least one ethylenically unsaturated carboxylic acid or its corresponding anhydride, and at least one styrenic monomer, and has a mean particle size diameter of less than or equal to about 100 nm.

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WATER-WHITENING RESISTANT LATEX EMULSION PRESSURE SENSITIVE ADHESIVE AND ITS PRODUCTION

CROSS-REFERENCE TO RELATED APPLICATIONS

None

STATEMENT REGARDING FEDERALLY SPONSORED RESEARCH

5 Not applicable.

BACKGROUND OF THE INVENTION

The present invention relates to pressure sensitive adhesives based on aqueous latex emulsions and processes for the preparation of the adhesives.

10 Pressure sensitive adhesives prepared according to the present invention have mean particle diameter sizes of less than or equal to about 100 nm and narrow particle size distributions. These pressure sensitive adhesives are particularly suitable for applications that require that the pressure sensitive adhesive maintain adhesion between the substrate and facestock when subjected to hot water spraying or

15 immersion. In addition, the adhesives exhibit resistance to water-whitening or "blush", often determined by a cold or ice water immersion test. Hot water adhesion is required in applications such as bottle labels where the bottles are subjected to hot water spraying in washing operations. In general, resistance to water-whitening is desirable anywhere a pressure sensitive adhesive with transparent facestock or substrate is

20 subjected to water or high humidity. Examples include labels on the sides of trucks, signs, and bottles.

Methods of providing water-whitening resistant latex emulsions for use in pressure sensitive adhesives are disclosed in the art. U.S. Patents 5,286,843 and 5,536,811 disclose a process for improving the water-whitening resistance of pressure

25 sensitive adhesives containing an aqueous latex emulsion and water soluble ions by removing the water soluble ions and adjusting the pH to at least about 6. The patents disclose that water-soluble ions may be removed by a number of techniques including centrifugation, dialysis, precipitation and deionization with ion exchange resins. The preferred method of removing the water-soluble ions is to contact the aqueous latex

30 emulsion, the formulated pressure sensitive adhesive containing the aqueous emulsion or both with an ion exchange resin.

International Application WO 97/11996 discloses a process for preparing hot water-whitening resistant latex emulsions useful in pressure sensitive adhesive compositions. The process involves copolymerizing a monomer mixture containing at

least one alkyl acrylate ester of an alcohol containing at least 4 carbon atoms, at least one polar co-monomer and at least one partially soluble co-monomer present in an amount of at least about 7 weight-%. Polymerization is carried out in the presence of at least one nonionic surfactant containing at least 8 moles of ethylene oxide and at

5 least one anionic surfactant containing up to about 10 moles of ethylene oxide. The polymerization product is neutralized to produce an emulsion having a pH greater than 7 and containing particles having a volume average particle size diameter up to about 165 nm. An electrolyte may be added subsequent to polymerization to stabilize opacity of a film cast from the emulsion.

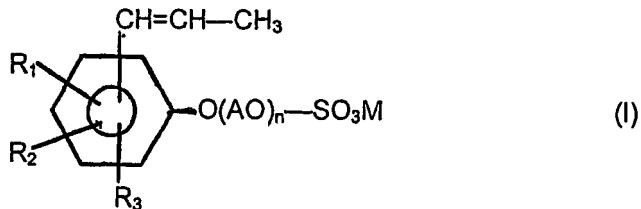
10 International Application WO 98/44064 discloses inherently tacky pressure sensitive adhesives prepared by emulsion polymerization of at least one monomer mixture comprising; at least one alkyl acrylate, the alkyl group of which has from 4 to 12 carbon atoms; at least one unsaturated carboxylic acid containing from about 3 to 5 carbon atoms and one styrenic monomer; wherein the particles have a mean diameter

15 of 300 nm or less. The publication discloses a single stage preparation of aqueous acrylic emulsions in examples 4D, 4E, 4F, 4G and 4H with average particle sizes ranging from 245 nm to 139 nm. Each of the examples discloses the use of silane crosslinkers to improve blush resistance. The publication discloses a preferred method of preparation, which yields adhesives resistant to water-whitening and involves a

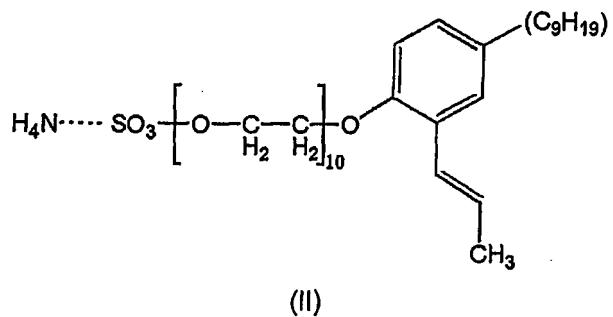
20 sequential polymerization of a first and second monomer charge. None of the above references disclose a pressure sensitive adhesive that maintains adhesion in hot water environments and is resistant to water-whitening.

BRIEF SUMMARY OF THE INVENTION

25 An aqueous, blush-retardant pressure sensitive adhesive (PSA) is made from an aqueous latex emulsion having an average particle size diameter of not substantially above about 100 nm and emulsified in the presence of an emulsifier consisting essentially of:



In structure I, R₁ is an alkyl, alkenyl, or aralkyl group containing between 6 and 18 carbon atoms, R₂ is H or R₁, R₃ is H or a propenyl group; A is an alkylene group of 2 to 4 carbon atoms, N is an integer ranging from 1 to 200, and M is an alkali metal, an 10 ammonium ion, or an alkanolamine cation. Preferably, the emulsifier is:



15 The preferred aqueous latex emulsion is prepared from a monomer mixture consisting essentially of at least one alkylacrylate having at least 4 carbon atoms in the alkyl chain, at least one ethylenically unsaturated carboxylic acid or its corresponding anhydride, and at least one styrenic monomer, and has a mean particle size diameter of less than or equal to about 100 nm.

20 Pressure sensitive adhesives of the present invention are useful in clear label applications, marking films, etc. The inventive pressure sensitive adhesives maintain adhesion and transparency (water-whitening resistance) when immersed in boiling water for 10 minutes. The inventive PSA's also possess good wet-out.

DETAILED DESCRIPTION OF THE INVENTION

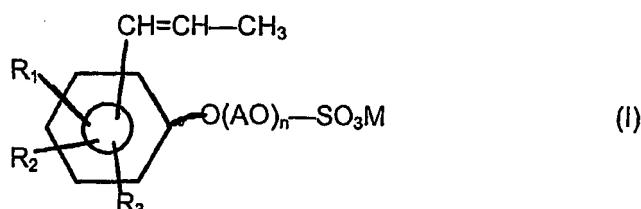
25 The polymerization is carried out in the presence of a reactive emulsifier or surfactant as described below. A redox type free radical initiator system is used in an amount sufficient to promote free radical polymerization of the monomers. Once the polymerization is complete it may be desirable to adjust the pH of the latex emulsion in order to enhance its stability. Other ingredients commonly used in the preparation of 30 aqueous latex emulsions such as buffering agents, chain transfer agents, and the like may be present. General latex technology is discussed in, Kirk-Othmer, *Encyclopedia of Technology*, [4thEd.], vol.15, p.51-65; which is hereby incorporated by reference. In addition to the aqueous latex emulsion, the pressure sensitive adhesive may also

contain additional components such as, biocides, wetting agents, defoamers, tackifiers, etc.

The reactive surfactant or emulsifier consists essentially of a compound represented by the following general structure:

5

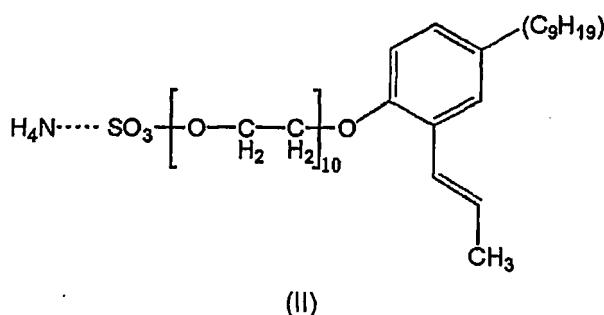
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In structure I, R₁ is an alkyl, alkenyl, or aralkyl group containing between 6 and 18 carbon atoms, R₂ is H or R₁, R₃ is H or a propenyl group; A is an alkylene group of 2 to 4 carbon atoms, N is an integer ranging from 1 to 200, and M is an alkali metal, an ammonium ion, or an alkanolamine cation. This reactive emulsifier can be made in accordance with the procedure described in U.S. Patent No. 5,332,854, the disclosure of which is expressly incorporated herein by reference.

Preferably, the emulsifier has the following general structure:

20



25

This preferred emulsifier is commercially available from Montello (Tulsa, OK) as Hitenol BC-10, poly(oxy-1,2-ethanediyl), α -sulfo- ω -[4-nonyl-2-(1-propenyl)phenoxy]-branched ammonium salts; yellowish brownish viscous liquid, 97.0 % actives, combined sulfuric acid content of 8.70-9.70%, pH of 6.5-8.5 (1% aqueous solution). The pendant double bond is reactive in the latex emulsion preparation if a slightly higher temperature is used, higher amounts of redox catalyst are employed, and a hydrophilic monomer is included in the monomer mix. Typically, less than 4 wt-% of the reactive surfactant based on the total weight of the latex, is used.

30

While use of a reactive emulsifier can be employed with a variety of latex emulsions for formulating the novel PSA's, a preferred latex emulsion is disclosed in commonly-assigned application serial no. 09/290,159, filed on April 12, 1999. The monomers used to prepare such aqueous lattices include alkyl acrylates, ethylenically

unsaturated carboxylic acids and their corresponding anhydrides and styrenic monomers.

Alkyl acrylates are alkyl esters of acrylic or methacrylic acid having at least 4 carbon atoms in the alkyl portion of the molecule. Examples include butyl acrylate, 5 isobutyl acrylate, heptyl acrylate, octyl acrylate; isoctyl acrylate, 2-ethylhexyl acrylate, and isodecyl acrylate. A single alkyl acrylate or mixtures of more than one alkyl acrylate can be used. A preferred alkyl acrylate is 2-ethylhexyl acrylate. The alkyl acrylate monomers are present in the monomer mixture in an amount from about 50 wt-% to about 90 wt-% and more preferably from about 60 wt-% to about 65 wt-% 10 based on the total weight of the monomer mixture.

Examples of ethylenically unsaturated carboxylic acids and their corresponding anhydrides used in the present invention include acrylic acid, methacrylic acid, maleic acid, fumaric acid, itaconic acid, beta-carboxyethyl acrylate and maleic anhydride. A single ethylenically unsaturated carboxylic acid or its corresponding anhydride or 15 mixtures thereof can be used. A preferred carboxylic acid is beta-carboxyethyl acrylate. The ethylenically unsaturated carboxylic acids or their corresponding anhydrides are present in the monomer mixture in amounts from about 5 wt-% to about 10 wt-%, more preferably from about 6 wt-% to about 8 wt-% based on the total weight of the monomer mixture.

20 Examples of styrenic monomers used in the present invention include styrene, t-butyl styrene, dimethyl styrene, and vinyl toluene. A preferred monomer is styrene. Some or all of the styrenic monomer content can be replaced with a hard monomer, *i.e.*, a monomer having a $T_g > 30^\circ C$. Representative hard monomers include, *inter alia*, methyl methacrylate (MMA), isobornyl acrylate, vinyl acetate, and the like. The 25 styrenic monomers are present in the monomer mixture in amounts ranging from about 15 wt-% to about 40 wt-%, and advantageously from about 28 wt-% to about 34 wt-%, based on the total weight of the monomer mixture.

A redox type free radical initiator system is used to promote polymerization of the monomers. The initiator is peroxide or hydroperoxide such as t-butyl 30 hydroperoxide. The reducing agent used in the redox system is zinc formaldehyde sulfoxylate, sodium formaldehyde sulfoxylate, ascorbic acid, isoascorbic acid, sodium metabolisulfite and the like. A preferred redox type system consists of t-butyl hydroperoxide and zinc formaldehyde sulfoxylate.

The aqueous latex emulsions, which form the basis of the pressure sensitive 35 adhesives of the present invention, are prepared in a single stage synthesis with or without a seed in the reaction vessel prior to beginning the monomer feed. Reaction temperatures during the monomer feed can range from about $50^\circ C$ to about $90^\circ C$. In

a preferred method of preparing the aqueous latex a pre-emulsion, an aqueous solution of the initiator, and an aqueous solution of a reducing agent are prepared in separate vessels. A reaction vessel is charged with deionized water, an anionic surfactant and a predetermined amount of initiator. The mixture in the reaction vessel

5 is heated with stirring and up to 20 wt % of the pre-emulsion, more preferably up to 8 wt-% and most preferably 4 wt-% is added to the reaction vessel along with a predetermined amount of the reducing agent to form the seed. In small batches such as laboratory size synthesis the predetermined amount of reducing agent, the "initial Zn hit", can be added before the initial pre-emulsion charge. In larger scale synthesis

10 it is preferred that the predetermined amount of reducing agent is added after the initial pre-emulsion charge is added to the reaction vessel. After forming the seed, the contents of the reaction vessel are heated to a desired temperature and the pre-emulsion, initiator and the reducing agent are simultaneously metered into the reaction vessel with stirring. It has been found to be advantageous to mix the pre-emulsion

15 and initiator. This can be accomplished by merging the pre-emulsion and initiator feed streams and passing the merged stream through a static mixer or by simply allowing the two feed streams to converge in a common feed line. On completion of the pre-emulsion feed, the contents of the reaction vessel are cooled and alternating predetermined amounts of the initiator and reducing agent are added to the reaction

20 vessel with stirring. This alternating initiator/reducing agent addition is preferably performed at least once. Once the reaction is complete the pH may be adjusted. The pH of the aqueous latex emulsion is preferably adjusted to a pH of about 6 to about 9 and more preferably about 6 to about 7.5. For efficiency and economy an aqueous solution of ammonium hydroxide can be used to adjust the pH. Other bases that may

25 be used include amines, imines, alkali metal and alkaline metal hydroxides, carbonates, etc.

In addition to the aqueous latex emulsion, the pressure sensitive adhesive composition advantageously contains biocides, wetting agents, defoamers, etc. Examples of suitable biocides include Kathon LX, commercially available as a 1.5% solution from Rohm & Haas and Metatin 910, commercially available from ACIMA. An example of a suitable wetting agent is Surfynol SE commercially available from Air Products, PLURONIC® type polyols commercially available from BASF Corp, and the like. Examples of defoamers include Drewplus T-1201 and Drewplus 1-191 commercially available from Ashland Specialty Chemical Company, and Rhodoline 6681, commercially available from Rhodia.

The pressure sensitive adhesives described above can be used to prepare articles such as tapes, labels, signs, marking films, and the like. In a typical

construction the pressure sensitive adhesive is coated or otherwise applied to a release liner such as a siliconized paper, dried, and laminated to a facestock. Alternatively, the pressure sensitive adhesive is coated directly on a facestock. Examples of facestocks include cellulosics, metal foils, polycarbonates, polyethylene, 5 polypropylene, polyethylene terephthalate, and vinyl films.

The pressure sensitive adhesives typically have a viscosity after adjusting the pH to between about 6 and about 8 of from about 1,000 to about 20,000 centipoises. The pressure sensitive adhesives exhibit a shear-thinning rheology such that it allows coating even on difficult to coat films. Conventional coating techniques can be used to 10 apply the pressure sensitive adhesives. Such techniques include dipping, slot die, air knife, brush curtain, extrusion blade, reverse roll, squeeze roll coating, and the like.

While the invention has been described with reference to preferred embodiments, those skilled in the art will understand that various changes may be made and equivalents may be substituted for elements thereof without departing from 15 the scope of the invention. In addition, many modifications may be made to adapt a particular situation or material to the teachings of the invention without departing from the essential scope thereof. Therefore, it is intended that the invention not be limited to the particular embodiment disclosed as the best mode contemplated for carrying out this invention, but that the invention will include all embodiments falling within the 20 scope of the appended claims. In this application all units are in the metric system and all amounts and percentages are by weight, unless otherwise expressly indicated. Also, all citations referred herein are expressly incorporated herein by reference.

IN THE EXAMPLES

The following test procedures were used in the examples:

1. 180° Peel Test: PSTC-1 (November 1975), Pressure Sensitive Tape Council, Glenview, Ill. Results of this test are reported in pounds/inch for a 1 in strip.
- 5 2. 178° Shear Test: Modified PSTC-7 using 1 x 1 x 4 lbs (November 1975). Pressure Sensitive Tape Council. Results of this test are reported in hours/500 gm/0.25 in² at 22°C.
- 10 3. Polyken Tack Test: This test is conducted on a Polyken, Jr. Probe Tack Tester (Polyken is a trademark of the Kendall Company) supplied by Testing Machines, Inc. (Amityville, N.Y.) under the following conditions:
Probe: 304 SS. 0.5 cm. diameter probe with a 280 grit abrasive finish.
Dwell Time: 1 second
15 Probe Contact Pressure: 100 gm/cm²
Probe Retraction Rate: 1 cm/sec.
Annular Weight: 20 gm. - 100 gm/cm² pressure of a 0.5 cm. diameter probe
Procedure: A one-inch square of MYLAR polyester film coated with the adhesive is placed on top of the annular weight so that the hole is completely covered by the adhesive area and this assembly placed in the weight carrier well. The machine is activated and the sequence of probe pressure and probe retraction automatically accomplished. The force required to free the probe from adhesive coated film, measured in grams/cm² is read from the Indicator dial on the machine.
- 20 4. Adhesive Failure Code:
C = clean, adhesive failure
gh = ghosting
PT = partial transfer
TT = total transfer
- 25 30 35

EXAMPLE 1

A typical formulation of the invention PSA is as follows:

TABLE 1

Composition of 6448-79 Latex

Component	Wt-% Based on Latex
Water	51.80
Sodium bicarbonate	0.10
Hitenol BC-10*	1.12
70% <i>t</i> -Butyl hydroperoxide	0.19
2-Ethyl hexylacrylate	32.27
Styrene	7.61
Methyl methacrylate	3.81
β -carboxyethyl acrylate	2.48
Methacrylic acid	1.50
Zinc formaldehyde sulfoxylate	0.12

*Hitenol BC-10 is poly(oxy-1,2-ethanediyl), α -sulfo- ω -[4-nonyl-2-(1-propenyl)phenoxy]-branched ammonium salts; yellowish brownish viscous liquid, 97.0 % actives, combined sulfuric acid content of 8.70-9.70%, pH of 6.5-8.5 (1% aqueous solution), supplied by Montello, Tulsa, OK.

A typical synthesis is set forth below.

10 Preparation of the Pre-Emulsion

To a 500 ml. pre-emulsion vessel equipped with a turbine agitator was charged de-ionized water (64.8 g.), NaHCO₃ (0.4 g.), 70% *t*-butyl hydroperoxide initiator (*t*-BHP, 0.60 g.), and Hitenol BC-10 polymerizable anionic surfactant (3.6 g.). The agitation was adjusted to 400 rpm. A monomer solution consisting of beta-carboxyethyl acrylate (B-CEA, 10.0 g.), methacrylic acid (MAA, 2.0 g.), methyl methacrylate (MMA, 15.33 g.), 2-ethylhexyl acrylate (2-EHA, 129.9 g.), and styrene (30.65 g.) then was slowly added to the vessel. Agitation of the emulsion was continued for 35 minutes after which the pre-emulsion was transferred to the reservoir of a metering pump system for eventual delivery to the polymerization reaction vessel.

20

Preparation of The Reducing Agent Feed Solution

A solution of zinc formaldehyde sulfoxylate (ZFS, 0.35 g.) in de-ionized water (12.0 g) was prepared and added to the reservoir of a peristaltic pump for eventual delivery to the polymerization reaction vessel.

25

Reactor charge and polymerization

To a 500 ml. reaction vessel equipped with a turbine agitator, thermocouple, heating mantle, temperature regulating device, N₂ sparge, and delivery lines for the pre-emulsion and reducing agent, was added de-ionized water (118.1 g.) and Hitenol

5 BC-10 polymerizable surfactant (0.90 g.). A N₂ sparge was started, the agitation set at 200 rpm, and the heating mantle was turned on. When the temperature reached 60° C, the N₂ sparge was turned off and 4% (~12 ml.) of the pre-emulsion was pumped into the reaction vessel. When the temperature reached 70° C, a single addition of ZFS reducing agent (0.07 g.) solution in de-ionized water (5.0 g.) was added.

10 Formation of a translucent blue dispersion within a few minutes indicated that the polymerization had initiated. Heating was continued to the controlled polymerization temperature of 80° C, whereupon the pre-emulsion and ZFS reducing agent feeds were started. The addition rates were adjusted to complete the deliveries over a three-hour time period at a reaction temperature of 80° C. Ten minutes after

15 completion of the feeds, additional initiator (70% t-BHP, 0.075 g.) in de-ionized water (1.25 g.) was added, followed after another 10 minutes reaction time by additional ZFS reducing agent (0.03 g.) in de-ionized water (1.25 g.). The reaction was held at 80° C for one additional hour after which cooling was started. When the temperature reached 50° C additional initiator (70% t-BHP, 0.075 g.) in de-ionized water (1.25 g.)

20 was added, followed after another 10 minutes by additional ZFS reducing agent (0.03 g.) in de-ionized water (1.25 g.). Cooling was continued to a temperature < 30° C at which point the latex was removed.

Additional formulations were compounded like latex 6448-79 and evaluated for its properties. The formulations evaluated and results recorded are set forth below:

TABLE 2
Run 6448-

	% BOM	73	75	79	81	85	87	91	93	95	54*
2-Ethyl hexylacrylate	58.6	58.6	69.15	69.15	65.55	68.36	73.71	72.51	72.51	60.37	
Styrene	0	16.31	16.31	8.16	16.31	8.16	0	0	0	0	30.72
Methyl methacrylate	32.62	16.31	8.16	16.31	8.16	16.31	16.31	16.31	16.31	—	
β -carboxyethyl acrylate	5.32	5.32	5.32	5.32	5.32	3.57	5.32	5.32	5.32	5.32	5.12
Methacrylic acid	1.06	1.06	1.06	1.06	1.06	0	1.06	1.06	1.06	1.06	1.02
Hitenol BC-10	2.4	2.4	2.4	2.4	3.6	3.6	3.6	4.8	4.8	4.8	2.77
<hr/>											
TEST RESULTS											
Shear (1 x 1 x 4 lb)	144+	144+	102.6 cf	104 cf	87+	15.9 cf	11.5 cf	12.4 cf	35.7 cf	35.7 cf	17.86
15 min. peel	2.3 cl	2.4 cl	2.2 cl	2.3 cl	2.32 cl	1.9 cl	1.5 cl	1.5 cl	1.5 cl	1.5 cl	2.6 cl
24 hr peel	3.0 cl	3.5 cl	3.2 cl	3.1 cl	3.0 cl	2.9 cl	2.8 cl	2.4 cl	2.3 cl	2.3 cl	3.9 cl
Polyken tack test	40	67	183	107	151	197	203	242	218	218	470
Particle Size (nm)	100	86.1	86	88	81.8	81.5	92	86.4	97.6	97.6	82.1
Blush (days) (tap water immersion)	19+	16+	2	14+	10+	6	7+	4+	3+	3+	

* Run 6497-54

cl = clean

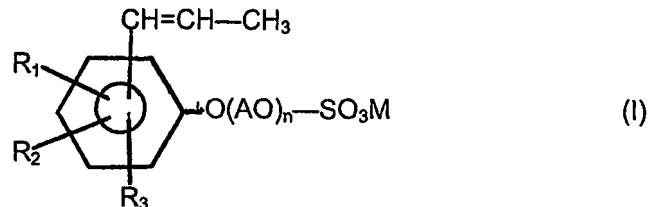
cf = cohesive failure

+ = greater than

The above-tabulated results demonstrate the remarkable properties exhibited by the inventive PSA's that utilize an aqueous latex emulsion PSA that employs a reactive emulsifier as the only emulsifier used to make the latex emulsion. Addition of non-reactive emulsifiers, while a small amount is tolerable, will degrade the otherwise excellent performance exhibited by the inventive PSA's. Note also that in test 54 no hard monomer (MMA) was used and the remarkable properties still were exhibited.

CLAIMS

1. An aqueous, blush-retardant pressure sensitive adhesive (PSA), which comprises:
 an aqueous latex emulsion having an average particle size diameter of not substantially above about 100 nm and emulsified in the presence of an
 5 emulsifier consisting essentially of:



10

wherein,

R₁ is an alkyl, alkenyl, or aralkyl group containing between 6 and 18 carbon atoms,

15

R₂ is H or R₁,

R₃ is H or a propenyl group;

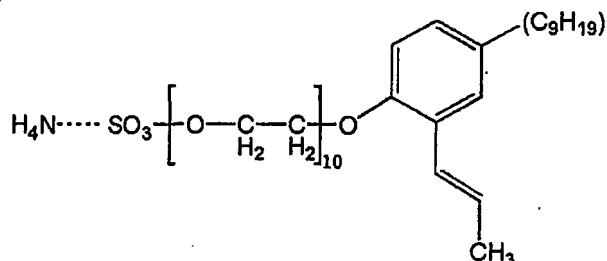
A is an alkylene group of 2 to 4 carbon atoms,

N is an integer ranging from 1 to 200, and

M is an alkali metal, an ammonium ion, or an alkanolamine cation.

20

2. The PSA of claim 2, wherein said emulsifier is represented by the following structure:



(II)

25

3. The PSA of claim 1, wherein said latex emulsion is prepared from a monomer mixture consisting essential of:

- at least one alkyl acrylate having at least 4 carbon atoms in the alkyl group,
- 30 at least one ethylenically unsaturated carboxylic acid or its corresponding anhydride, and
- at least one styrenic monomer,

in the presence of said emulsifier and a redox type free radical initiator system.

4. The PSA of claim 3, wherein the alkyl group of said alkyl acrylate having at least 4 carbon atoms in the alkyl group is one or more of butyl acrylate, 2-ethylhexyl acrylate, heptyl acrylate, octyl acrylate, isoctyl acrylate, isobutyl acrylate, or isodecyl acrylate.

5. The PSA of claim 3, where the ethylenically unsaturated carboxylic acid or its corresponding anhydride is one or more of acrylic acid, methacrylic acid, beta-carboxyethyl acrylate, maleic acid, fumaric acid, itaconic acid, or maleic anhydride.

10 6. The PSA of claim 3, wherein the styrenic monomer is one or more of styrene, tert-butyl styrene, dimethyl styrene, or vinyl toluene.

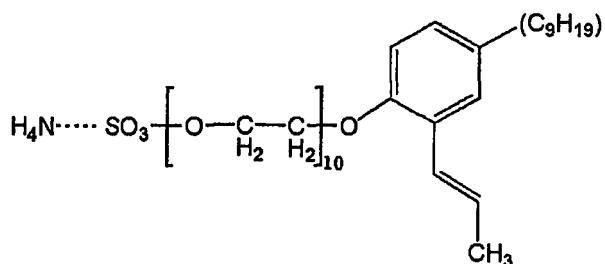
15 7. The PSA of claim 3, wherein the styrenic monomer is at least partially replaced with a hard monomer having a $T_g > 30^\circ C$.

20 8. The PSA of claim 7, wherein said hard monomer is one or more of methyl methacrylate, isobornyl acrylate, and vinyl acetate.

9. The PSA of claim 1, which further comprises a wetting agent.

25 10. An aqueous, blush-retardant pressure sensitive adhesive (PSA), which comprises:
an aqueous latex emulsion having an average particle size diameter of not substantially above about 100 nm and emulsified in the presence of an emulsifier consisting essentially of:

30



(II)

wherein said latex emulsion is made from a monomer mixture that includes beta-carboxyethyl acrylate

11. The PSA of claim 10, wherein said monomer mixture further consists
5 essential of:
 - a) at least one alkyl acrylate having at least 4 carbon atoms in the alkyl group,
 - b) at least one ethylenically unsaturated carboxylic acid or its corresponding anhydride, and
 - 10 c) at least one styrenic monomer,
in the presence of said emulsifier and a redox type free radical initiator system.
12. The pressure sensitive adhesive of claim 9, wherein the alkyl group of said
15 alkyl acrylate having at least 4 carbon atoms in the alkyl group is one or more of butyl acrylate, 2-ethylhexyl acrylate, heptyl acrylate, octyl acrylate, isoctyl acrylate, isobutyl acrylate, or isodecyl acrylate.
13. The pressure sensitive adhesive of claim 11, where the ethylenically
20 unsaturated carboxylic acid or its corresponding anhydride is one or more of acrylic acid, methacrylic acid, beta-carboxyethyl acrylate, maleic acid, fumaric acid, itaconic acid, or maleic anhydride.
14. The pressure sensitive adhesive of claim 11, wherein the styrenic monomer is
25 one or more of styrene, tert-butyl styrene, dimethyl styrene, or vinyl toluene.
15. The PSA of claim 11, wherein the styrenic monomer is at least partially replaced with a hard monomer having a $T_g > 30^\circ C$.
- 30 16. The PSA of claim 15, wherein said hard monomer is one or more of methyl methacrylate, isobornyl acrylate, and vinyl acetate.
17. A laminate comprising:
 - (a) a release liner,
 - 35 (b) a facestock, and
 - (c) the PSA of claim 1 between and adhering the release liner and facestock.

18. A laminate comprising:

- (a) a release liner,
- (b) a facestock, and
- (c) the PSA of claim 2 between and adhering the release liner and facestock.

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19. A laminate comprising:

- (a) a release liner,
- (b) a facestock, and

10 (c) the PSA of claim 3 between and adhering the release liner and facestock.

20. 14. A laminate comprising:

- (a) a release liner,
- 15 (b) a facestock, and
- (c) the PSA of claim 4 between and adhering the release liner and facestock.

21. 14. A laminate comprising:

20 (a) a release liner,

- (b) a facestock, and
- (c) the PSA of claim 5 between and adhering the release liner and facestock.

25 22. A laminate comprising:

- (a) a release liner,
- (b) a facestock, and
- (c) the PSA of claim 6 between and adhering the release liner and facestock.

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23. A laminate comprising:

- (a) a release liner,
- (b) a facestock, and
- (c) the PSA of claim 7 between and adhering the release liner and facestock.

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24. A laminate comprising:

- (a) a release liner,
- (b) a facestock, and
- (c) the PSA of claim 8 between and adhering the release liner and facestock.

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25. A laminate comprising:

- (a) a release liner,
- (b) a facestock, and
- (c) the PSA of claim 9 between and adhering the release liner and facestock.

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26. A tape comprising:
a wound roll of facestock having an exterior side and an interior side, the interior side coated with the PSA of claim 1.

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27. A tape comprising:
a wound roll of facestock having an exterior side and an interior side, the interior side coated with the PSA of claim 2.

20

28. A tape comprising:
a wound roll of facestock having an exterior side and an interior side, the interior side coated with the PSA of claim 3.

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29. A tape comprising:
a wound roll of facestock having an exterior side and an interior side, the interior side coated with the PSA of claim 4.

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30. A tape comprising:
a wound roll of facestock having an exterior side and an interior side, the interior side coated with the PSA of claim 5.

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31. A tape comprising:
a wound roll of facestock having an exterior side and an interior side, the interior side coated with the PSA of claim 6.

32. A tape comprising:

a wound roll of facestock having an exterior side and an interior side, the interior side coated with the PSA of claim 7.

33. A tape comprising:

5 a wound roll of facestock having an exterior side and an interior side, the interior side coated with the PSA of claim 8.

34. A tape comprising:

10 a wound roll of facestock having an exterior side and an interior side, the interior side coated with the PSA of claim 9.

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US01/14740

A. CLASSIFICATION OF SUBJECT MATTER

IPC(7) :Please See Extra Sheet.

US CL :Please See Extra Sheet.

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 524/547, 558, 560, 561, 577, 817, 824, 833; 526/287, 318, 318.4, 318.45, 911

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

BRS, CAS ONLINE

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 5,663,241 A (TAKAMATSU et al.) 02 September 1997, col. 3, lines 8-12 and Example 13.	1, 2, 17, 18, 26 and 27
Y		-----
A		9, 25 and 34
X	US 5,885,708 A (LU et al.) 23 March 1999, col. 5, lines 8-53 and col. 6, lines 18-46.	3-6, 10-14, 19-22 and 28-31
Y		1, 2, 17, 18, 26 and 27
A		-----
		9, 25 and 34

		3-6, 10-14, 19-22 and 28-31

Further documents are listed in the continuation of Box C. See patent family annex.

"A"	Special categories of cited documents:	"T"	later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"B"	document defining the general state of the art which is not considered to be of particular relevance	"X"	document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"C"	earlier document published on or after the international filing date	"Y"	document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
"D"	document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"Z"	document member of the same patent family
"E"	document referring to an oral disclosure, use, exhibition or other means		
"F"	document published prior to the international filing date but later than the priority date claimed		

Date of the actual completion of the international search

26 JULY 2001

Date of mailing of the international search report

27 AUG 2001

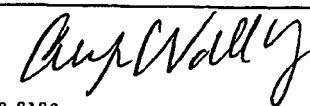
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INTERNATIONAL SEARCH REPORT

International application No.
PCT/US01/14740

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 5,620,796 A (KAWABATA et al.) 15 April 1997, col. 2, lines 1-15, col. 3, line 2, and col. 4, lines 1-50 and col. 5, lines 2-20.	1-6, 17-22 and 26-31
--Y		-----
--A		9, 25 and 34

X	EP 0 845 518 A1 (KAO CORPORATION) 06 March 1998, page 3, lines 27-31, page 6, lines 14-23 and 49, and page 8, lines 45-55.	1-6, 17-22 and 26-31
--Y		-----
--A		9, 25 and 34

X	US 5,332,854 A (YOKOTA et al.) 26 July 1994, col. 2, lines 12-66, col. 3, lines 12-59 and Example 1.	1, 2, 17, 18, 26 and 27
--Y		-----
--A		3-6, 9, 19-22, 25, 28-31 and 34

X	US 5,889,105 A (SATSUMA et al.) 30 March 1999, col. 3, lines 35-67, col. 4, lines 1-15 and Example 1.	1, 17, and 26
--Y		-----
--A		3-6, 9, 19-22, 25, 28-31 and 34

		2, 10-14; 18 and 27

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US01/14740

Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This international report has not been established in respect of certain claims under Article 17(3)(a) for the following reasons:

1. Claims Nos.:
because they relate to subject matter not required to be searched by this Authority, namely:

2. Claims Nos.: 7, 8, 15, 16, 23, 24, 32 AND 33
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:

Please See Extra Sheet.

3. Claims Nos.:
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1. As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:

4. No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest



The additional search fees were accompanied by the applicant's protest.



No protest accompanied the payment of additional search fees.

INTERNATIONAL SEARCH REPORT

International application No.

PCT/US01/14740

A. CLASSIFICATION OF SUBJECT MATTER:
IPC (7):

C09J 125/14, 125/18, 129/10, 133/02, 133/08, 133/02, 133/14, 141/00

A. CLASSIFICATION OF SUBJECT MATTER:
US CL :

524/547, 558, 560, 561, 577, 817, 824, 833; 526/287, 518, 518.4, 518.45, 911

BOX I. OBSERVATIONS WHERE CLAIMS WERE FOUND UNSEARCHABLE

2. Where no meaningful search could be carried out, specifically:

These claims recite "[t]he PSA of claim 3, wherein the styrenic monomer is at least partially replaced . . ." However, claim 3 recites that the styrene monomer is an ESSENTIAL component in the monomer mixture. The term "AT LEAST partially replaced" seems to be claiming that the styrenic monomer may be replaced partially to completely. This is inconsistent with the definition in claim 3 that the styrenic monomer is an essential component in the monomer mixture.